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continued 11.

## 11 2 Practice Problems Continued Answers

Honors Chemistry 11-2 Practice problems Name:\_\_\_\_\_ See sample problem 2 on pg. 357 for the  
following problems: 1. Determine the mass of zinc chloride produced when 5.30 grams of zinc react  
with hydrochloric acid.  $11.0 \text{ g ZnCl}_2$  2. Determine the mass of lithium hydroxide produced when 0.38g  
of lithium nitride reacts with water according to the following equation:  $\text{Li}_3\text{N} + 3 \text{H}_2\text{O} \rightarrow \text{NH}_3 \dots$

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10<sup>24</sup> Practice Problems (continued) Date Class 1 If you burned  $6.10 \times 10^{24}$  molecules of  $\text{C}_2\text{H}_6$ , produces  
0.37 mol of ethane ( $\text{C}_2\text{H}_6$ ), what mass of ethane did you burn? cut' sog 18 HOJ many formul units are in  
5.  $\text{N}_2$  gas. What volume will that gas occupy at 1 mol  $\text{C}_2\text{H}_6$  25. A canister with a volume of 694 L contains  
how many moles of oxygen at 1 mol  $3\text{R} 10$

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2 11.  $x^2y^3 x^3y^4$  12.  $p^6 q^3 a^1 b^3$  Express each using rational exponents. 13.  $x^5y^6$  14.  $27x^{10}y^5$  15.

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144x6y10 16. 21 3 c7 17. 1024a3 18. 36a8b5 Express each using radicals. 19. 64 1 3 20. 22a 3 2b 5 2 21.  
s 2 3 t 1 3 v 3 2 22. y 3 2 23. x 2 5 y 3 5 24. (x6y3) 1 2 z 2 3 8 2 3 8 1 3 Practice NAME \_\_\_\_\_ DATE  
\_\_\_\_\_ PERIOD \_\_\_\_\_ 11-1

## Chapter 11 Resource Masters

This is a collection of worked general chemistry and introductory chemistry problems, listed in alphabetical order. Included are printable pdf chemistry worksheets so you can practice problems and then check your answers. You may also browse chemistry problems according to the type of problem.

## Practice Chemistry with Worked Chemistry Problems

Chapter 11 Answers (continued) 36 Answers Algebra 1 Chapter 11 Chapter Project Activity 1: Writing  
5280 ft; multiply 15 ft by ; Activity 2: Calculating A reasonable height would be less than 7 ft.

## Chapter 11 Answers - Poudre School District

Honors Chemistry 11-3 practice problems Name: \_\_\_\_\_ See sample problem 5 on pg. 368 for the following problems: 1. Identify the limiting reactant when 5.1 g of lithium reacts with 1.5 L of fluorine gas at STP to produce lithium fluoride.

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Chapter 11.3 Practice Problems EXPECTED SKILLS: Know how to compute the dot product of two vectors. Be able to use the dot product to find the angle between two vectors; and, in particular, be able to determine if two vectors are orthogonal. Know how to compute the direction cosines of a vector.

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## Chapter 11.3 Practice Problems - Drexel University

Practice Problems: Stoichiometry (Answer Key) Balance the following chemical reactions: a.  $2 \text{CO} + \text{O}_2 \rightarrow 2 \text{CO}_2$  b.  $2 \text{KNO}_3 \rightarrow 2 \text{KNO}_2 + \text{O}_2$  c.  $2 \text{O}_3 \rightarrow 3 \text{O}_2$  d.  $\text{NH}_4\text{NO}_3 \rightarrow \text{N}_2\text{O} + 2 \text{H}_2\text{O}$  e.  $4 \text{CH}_3\text{NH}_2 + 9 \text{O}_2 \rightarrow 4 \text{CO}_2 + 10 \text{H}_2\text{O} + 2 \text{N}_2$  f.  $\text{Cr}(\text{OH})_3 + 3 \text{HClO}_4 \rightarrow \text{Cr}(\text{ClO}_4)_3 + 3 \text{H}_2\text{O}$  Write the balanced chemical equations of each reaction:

## Practice Problems: Stoichiometry

2. Name the following compounds, determine the molar mass, determine how many O atoms are present in one molecule/formula unit, determine the grams of oxygen in 1.00 mole of the compound, and determine how many moles of O atoms in 8.35 grams of the compound. a)  $\text{K}_2\text{CO}_3$  b)  $\text{Ca}(\text{NO}_3)_2$  c)  $\text{Cu}_2\text{SO}_4$  d)  $\text{N}_2\text{O}$

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