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12.1: Everyday Stoichiometry Last updated; Save as PDF Page ID 53789; Everyday Stoichiometry; Summary; Contributors and Attributions; You are in charge of setting out the lab equipment for a chemistry experiment. If you have twenty students in the lab (and they will be working in teams of two) and the experiment calls for three beakers and two ...

12.1: Everyday Stoichiometry - Chemistry LibreTexts

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Chemistry Chapter 12 Stoichiometry Quiz

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Chapter 12 Stoichiometry Test Answer Key A In any stoichiometry problem, the first step is always to calculate the number of moles of each reactant present. In this case, we are given the mass of $\text{K}_2\text{Cr}_2\text{O}_7$ in 1 mL of solution, which we can use to calculate the number of moles of $\text{K}_2\text{Cr}_2\text{O}_7$ contained in 1 mL:

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