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7. As a group, play level 1 of the balancing equation game. Write down the strategies your group uses to balance chemical equations.

BALANCING CHEMICAL EQUATIONS

1 8. Start level 2 of the balancing equation game. Take turns in your

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group to balance the equations in the sim, using your strategies from Level 1, and adding new strategies as needed.

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As a group, play level 1 of the balancing equation game. Write down the strategies your group uses to balance chemical equations. Make the atoms on one side equal to the atoms on the other side. Some atoms need same coefficient while it could be

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Chemical Equations
Change for others. 8. Start level 2 of
the balancing equation game.

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C.8.D: The student is expected to use
the Law of Conservation of Mass to
write and balance chemical . Subjects:
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None of the students in either of the two public schools had ever been exposed to a guided-inquiry chemistry lab. I implemented the lab as supplemental instruction to expose students to the topics of green chemistry and biomimicry, topics not commonly covered in high school

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them real-world practice with units ...

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Chemical Equations equation game.
Take turns in your group to balance

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the equations in the sim, using your strategies from Level 1, and adding new strategies as needed. Each person should be in charge of balancing at least one

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Adapted from Chemistry: A Guided Inquiry, 2 nd ed. by Moog and Farrell (Wiley and Sons, Inc.) Balancing Chemical Equations Teacher Notes
Time Required: Estimated 20-30 minutes for Part I and 50 minutes for Part II. It is suggested that Part I and Part II be completed on consecutive

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days. Prerequisites: Chemical formulas
and nomenclature

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Equations. A chemical equation is a

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written expression of a chemical reaction; e.g., $2 \text{H}_2 + \text{O}_2 \rightarrow 2 \text{H}_2\text{O}$

Reactants are written on the left, and products are written on the right. In a balanced equation the total numbers of atoms of each kind on both

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stoichiometry unit when balancing equations are introduced. Requires one 55-minute class period. • Lesson 2: Acid-Base Neutralization Reactions This lesson should be implemented at the beginning of the solution ... AP

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