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Chapter 5 Electrons In Atoms Answer
Key: SECTION 5.1 MODELS OF THE
ATOM (pages 127–132) Chapter 5
Electrons in Atoms43 SECTION 5.1
MODELS OF THE ATOM (pages
127–132) This section summarizes
the development of atomic theory.

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principle. electron configuration.
principle energy level. A three-
dimensional region around the
nucleus of an atom that.... electrons
occupy orbitals of the lowest energy
levels first.

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SECTION 5.1 MODELS OF THE ATOM

(pages 127–132) This section summarizes the development of atomic theory. It also explains the significance of quantized energies of

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electrons as they relate to the quantum mechanical model of the atom.
The Development of Atomic Models
(pages 127–128) 1.

SECTION 5.1 MODELS OF THE ATOM
(pages 127–132)

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SECTION 5.2 ELECTRON

ARRANGEMENT IN ATOMS (pages

133136) This section shows you

Chapter 8 Periodic Properties of the

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5.1 Revising the Atomic Model >

- According to the Pauli exclusion principle, an atomic orbital may describe at most two electrons. • To

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occupy the same orbital, two electrons must have opposite spins; that is, the electron spins must be paired.

5.2 Electron Arrangement in Atoms >
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